



*Integrity ★ Service ★ Excellence*

# Organic Materials Chemistry

Date: 7 Mar 2013

Charles Lee  
Program Officer  
AFOSR/RTD

Air Force Research Laboratory

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# 2013 AFOSR SPRING REVIEW



NAME: **Charles Lee**

## BRIEF DESCRIPTION OF PORTFOLIO:

To exploit the uniqueness of **organic/polymeric** materials technologies for enabling future capabilities currently unavailable by discovering and improving their unique properties and processing characteristics

## LIST SUB-AREAS IN PORTFOLIO:

Photonic Polymers/Organics

Electronic Polymers/Organics

Novel Properties Polymers/Organics

NanoTechnology



# Organic Materials Chemistry

## Research Objective and Challenges



To exploit the unique properties of organic materials for enabling technologies

for enabling technologies

and

Challenges

- Discovery

- Control

- Balance

Approaches

- Molecular

- Process

- Structure

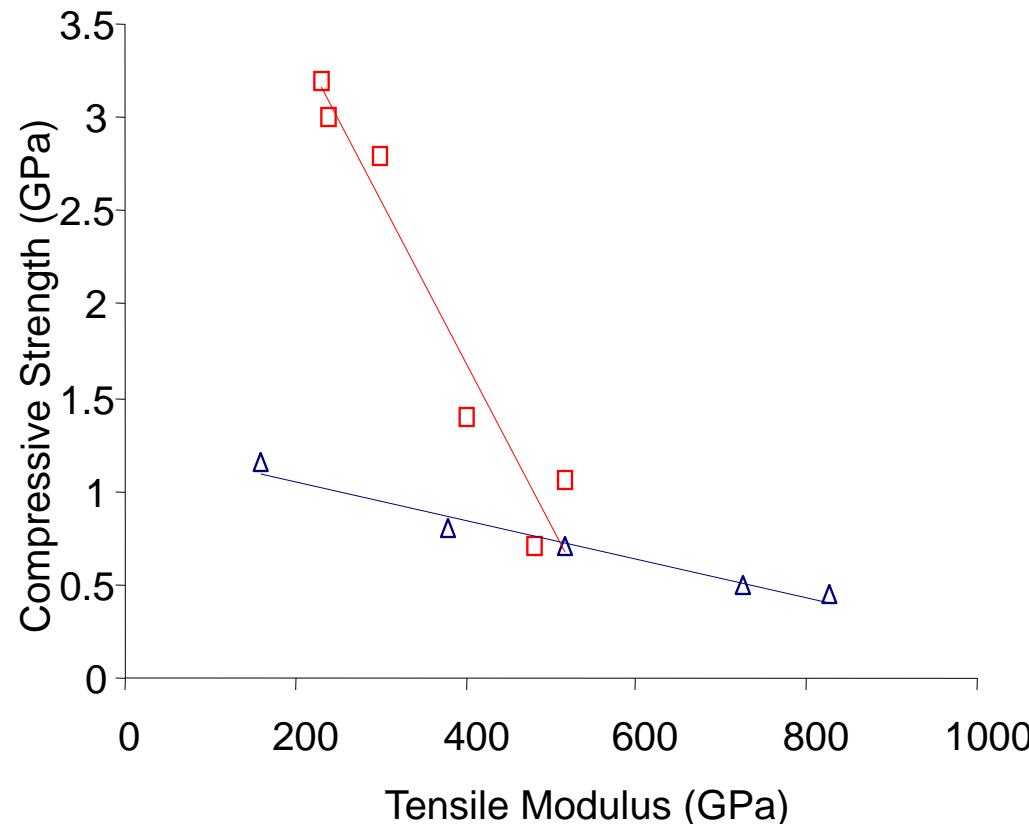
for enabling

technologies

and

enabling

technologies

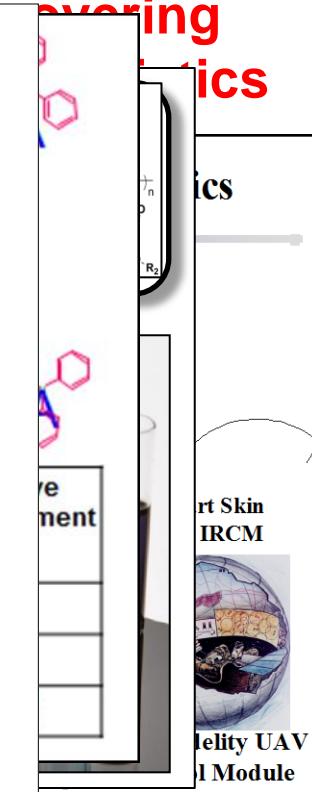


• Properties

• PAN Based Carbon Fibers

• Pitch Based Carbon Fibers

- Properties
- Not applications specific, but often use applications to guide the properties focuses

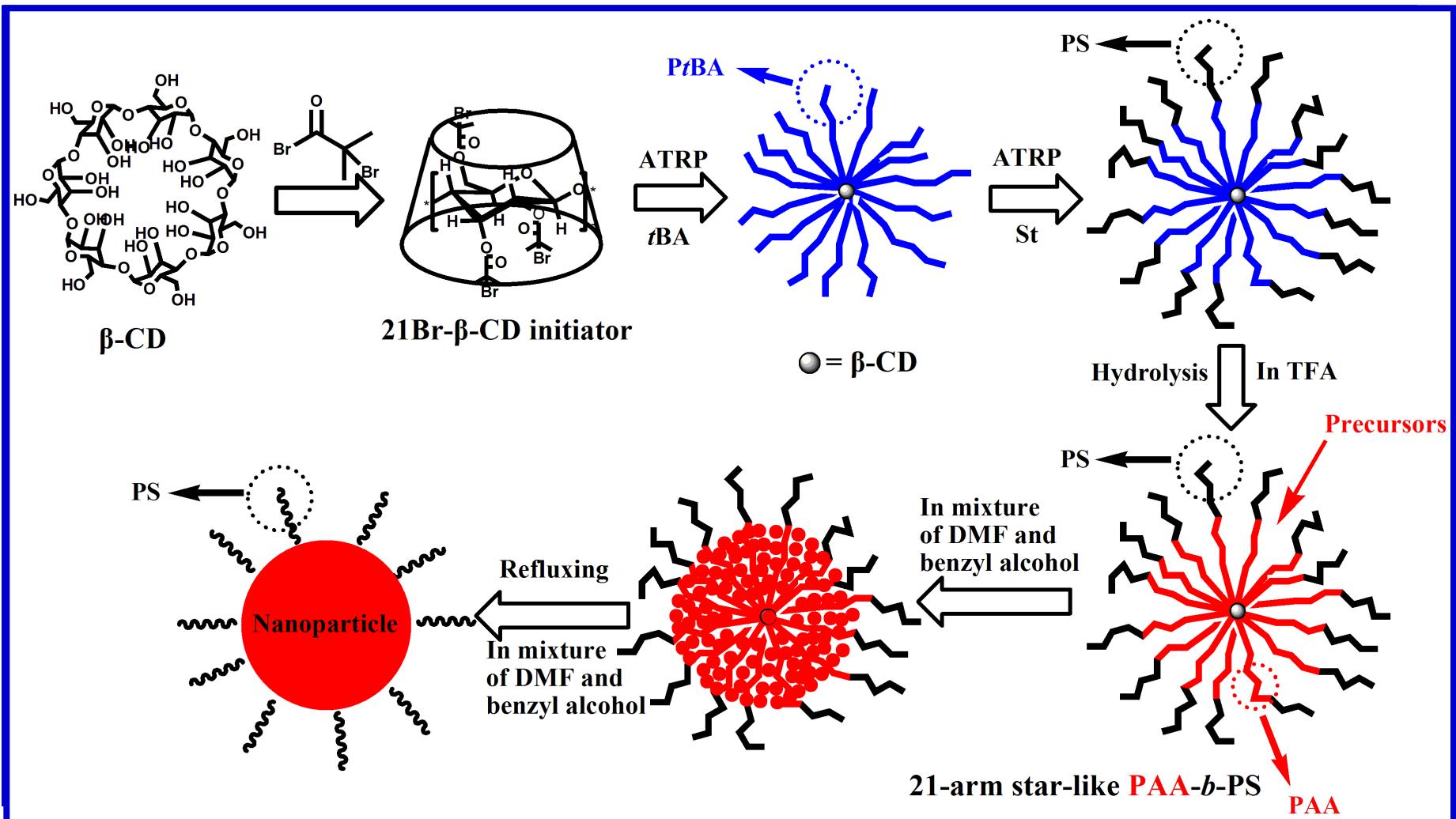


# Self Assembled Micelle vs Covalently Bonded

	Small Molecule	Block CoPolymer	Star-Like Molecule
Au – Diameter (nm)	<b>9±0.44</b>	<b>13±2</b>	<b>10.1±0.3</b>
Grams/L	<b>5.11</b>	<b>0.56</b>	<b>20.2</b>
# Particles/L	<b><math>6.9 \times 10^{17}</math></b>	<b><math>2.5 \times 10^{16}</math></b>	<b><math>2.0 \times 10^{18}</math></b>
Pt - Diameter (nm)	<b>73±5.74</b>	<b>6.0±0.98</b>	<b>6.2±0.2</b>
Grams/L	<b>4.86</b>	<b>0.86</b>	<b>26.3</b>
# Particles/L	<b><math>1.1 \times 10^{15}</math></b>	<b><math>3.6 \times 10^{17}</math></b>	<b><math>1.1 \times 10^{19}</math></b>
Fe <sub>2</sub> O <sub>3</sub> -Diameter (nm)	<b>16±1.49</b>	<b>10.8±2.98</b>	<b>10.1±0.5</b>
Grams/L	<b>2.94</b>	<b>1.81</b>	<b>36.2</b>
# Particles/L	<b><math>2.6 \times 10^{17}</math></b>	<b><math>6.552 \times 10^{17}</math></b>	<b><math>1.3 \times 10^{19}</math></b>
Cd-Se-Diameter (nm)	<b>8.5±0.65</b>	-----	<b>9.9±0.3</b>
Grams/L	<b>0.98</b>	-----	<b>22.8</b>
# Particles/L	<b><math>5.2 \times 10^{17}</math></b>	-----	<b><math>7.5 \times 10^{18}</math></b>
PbTiO <sub>3</sub> -Diameter (nm)	-----	<b>50±4.9</b>	<b>9.7±0.4</b>
Grams/L	-----	<b>2.12</b>	<b>31.2</b>
# Particles/L	-----	<b><math>4.1 \times 10^{15}</math></b>	<b><math>7.5 \times 10^{18}</math></b>

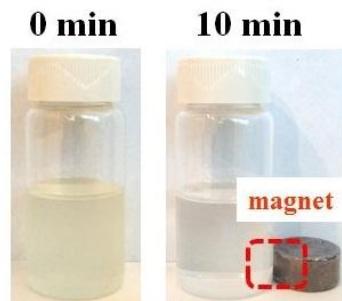
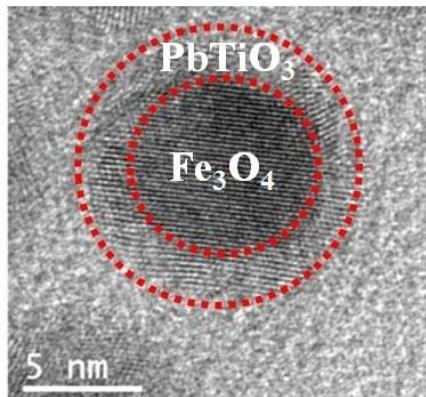
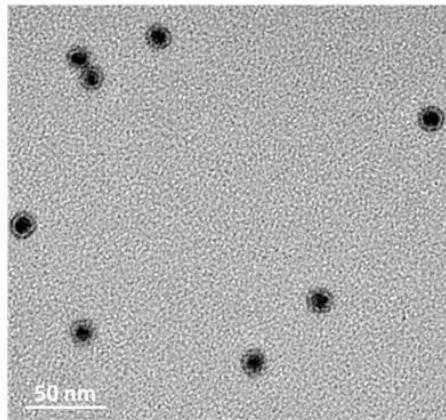
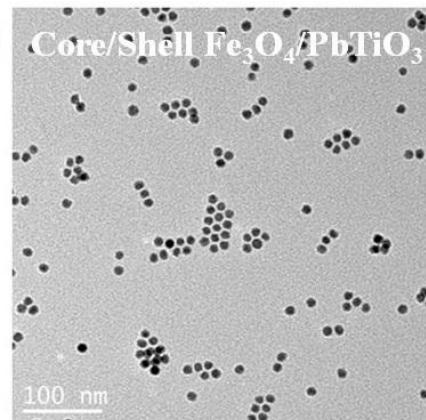


# NPs Synthesis by Novel Amphiphilic Star-Like Block Copolymers as Template





# Core/Shell Nanoparticles – with Large Lattice Mismatch



Core/shell nanostructures are *conventionally* obtained by dissimilar materials epitaxy, which requires moderate lattice mismatches (<2%) between the two different materials in order to obtain high-quality core/shell heterostructures, which would otherwise be difficult to obtain.

## $\text{Fe}_3\text{O}_4/\text{PbTiO}_3$

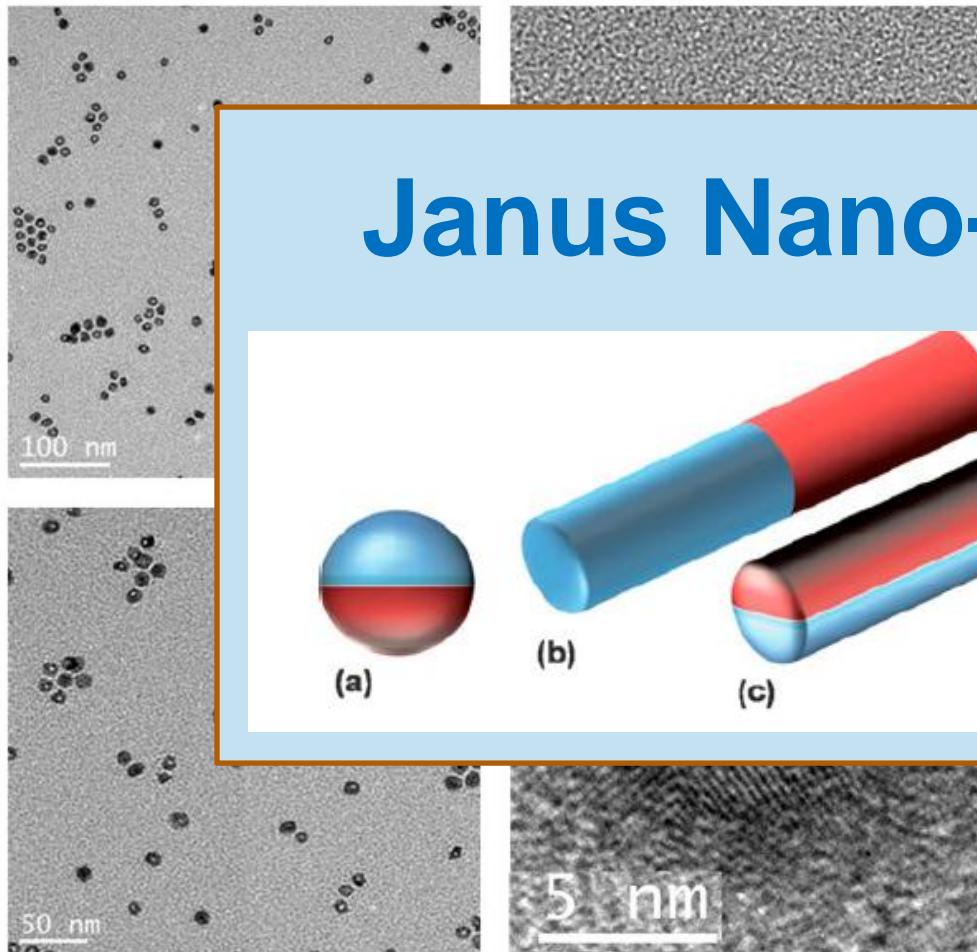
$$D_{\text{Fe}_3\text{O}_4} = 6.1 \pm 0.3 \text{ nm (core)}$$

$$D_{\text{PbTiO}_3} = 3.1 \pm 0.3 \text{ nm (shell)}$$

➤ Despite more than 40% lattice mismatch between  $\text{Fe}_3\text{O}_4$  and  $\text{PbTiO}_3$ ,  $\text{Fe}_3\text{O}_4/\text{PbTiO}_3$  core/shell nanoparticles can be readily crafted by this approach!!!

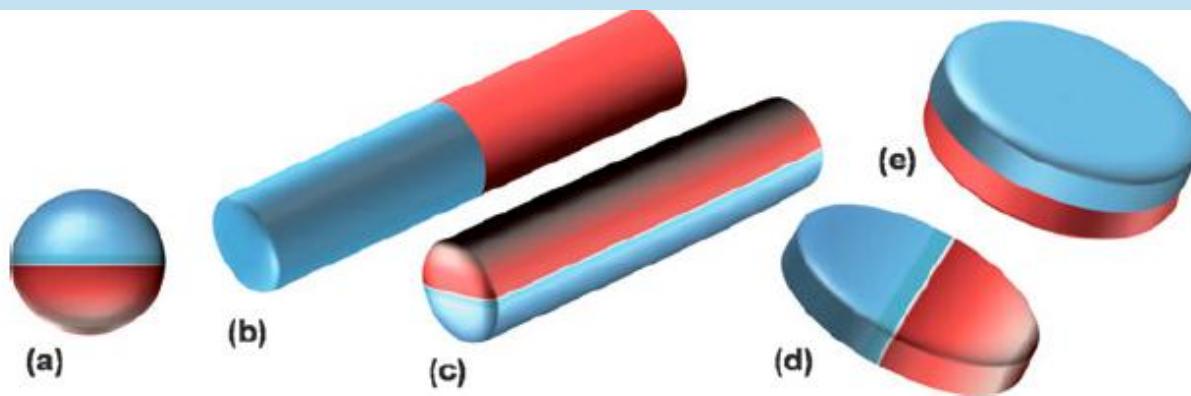


# Hollow Nanoparticles – Au Nanoparticles



Hollow noble metal

## Janus Nano-Particles



The diameter of hollow core  
 $= 5.6 \pm 0.4 \text{ nm}$



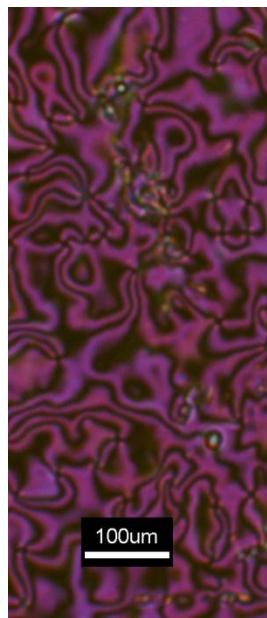
# Phototropic liquid crystals

Tim White, Tim Bunning, AFRL/RX



**“Phototropism”:** A term used to describe light induced phase changes in liquid crystals.

An example of



**Scheme for Light Induced Order-Disorder in Azobenzene Liquid Crystals**

Ikeda, J. Photochem. Photobio., 2004.

DISTRIBUTION STATEMENT A - Unclassified, Unlimited Distribution

AFRL 3

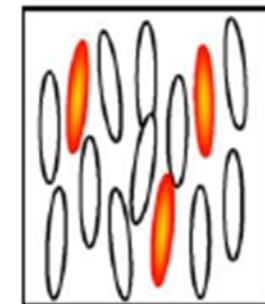
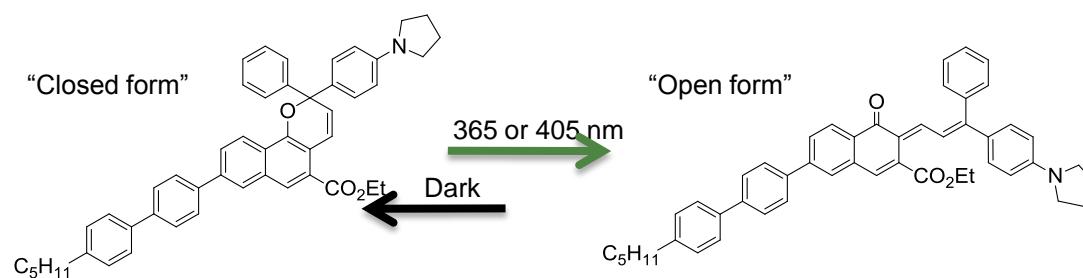
“Negative” phototropism – S (order parameter) decreases with light



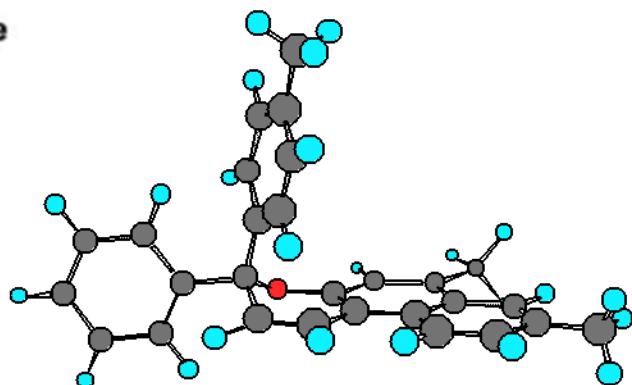
# Light Induced Disorder-Order in Napthopyran (AMI15)/LC Mixtures



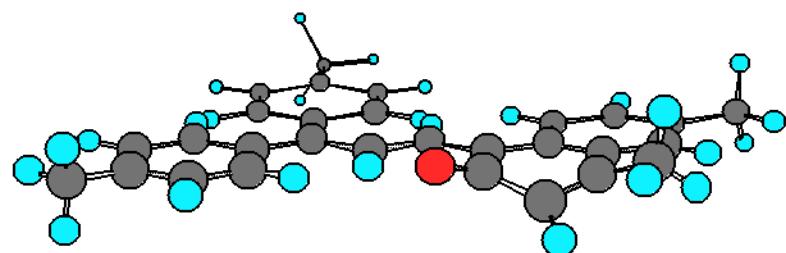
- New class of photochromic molecules that increase order upon light exposure employed for *disorder-order transitions*.
- *Demonstration of full gamut of Light Induced Phase Transitions*



I phase



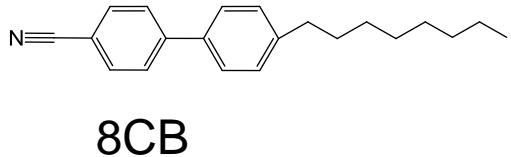
N phase



In this case of “positive” phototropism, illumination increases the compatibility of the napthopyran as the molecular shape becomes planar and quasi-rod like aligning favorably with the liquid crystalline phases.

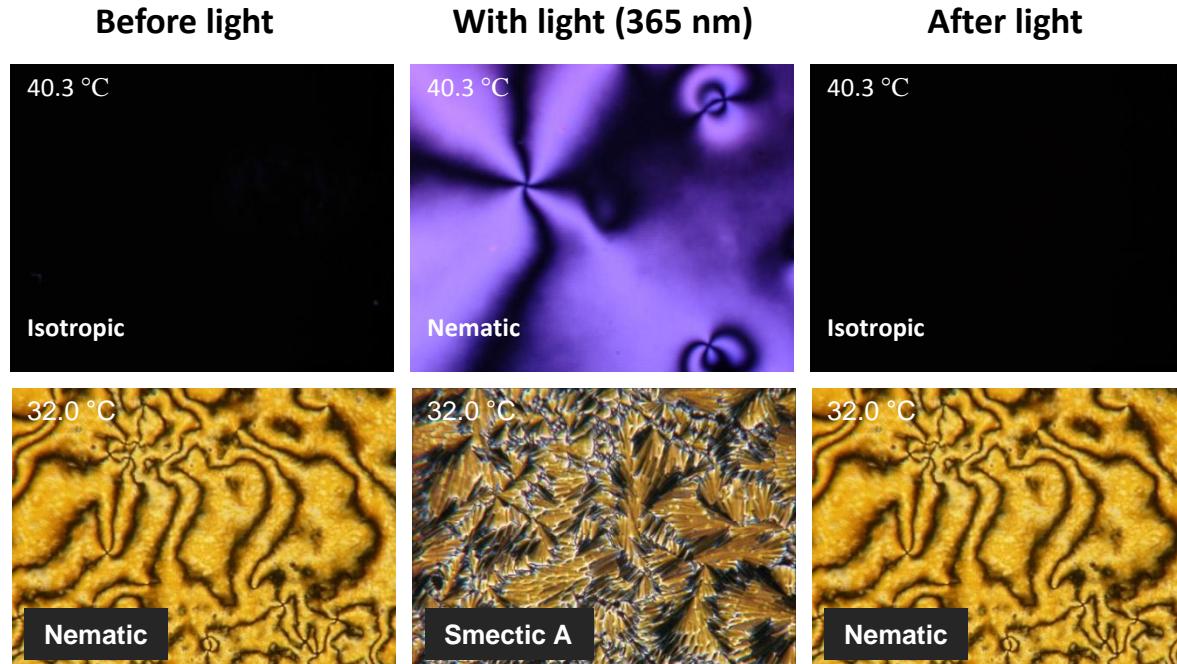


# AMI15/8CB Mixture Shows Additional Transition



**AMI15/8CB shows Photoinduced**

- Isotropic to Nematic Transition
- Nematic to Smectic A Transition



“Positive” phototropism – S (order parameter) increases with light

T. Kosa, L. Sukhomlinova, L. Su, B. Taheri, T.J. White, and T.J. Bunning, “Light Induced Liquid Crystallinity”, Nature, 2012, 485, 347-349.

DISTRIBUTION STATEMENT A – Unclassified, Unlimited Distribution



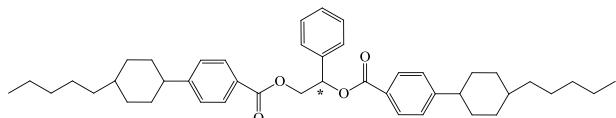


# Different Phase Change with Chiral Dopant

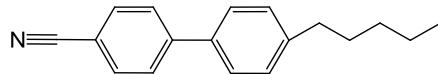


AMI15/5CB/R1011 Mixture shows Photoinduced:

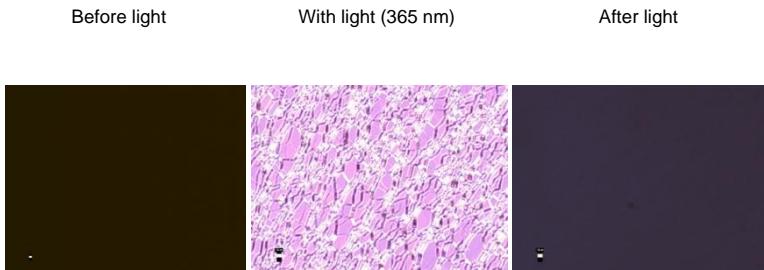
- Isotropic to Cholesteric Phase Transition



R1011 – a chiral dopant from Merck

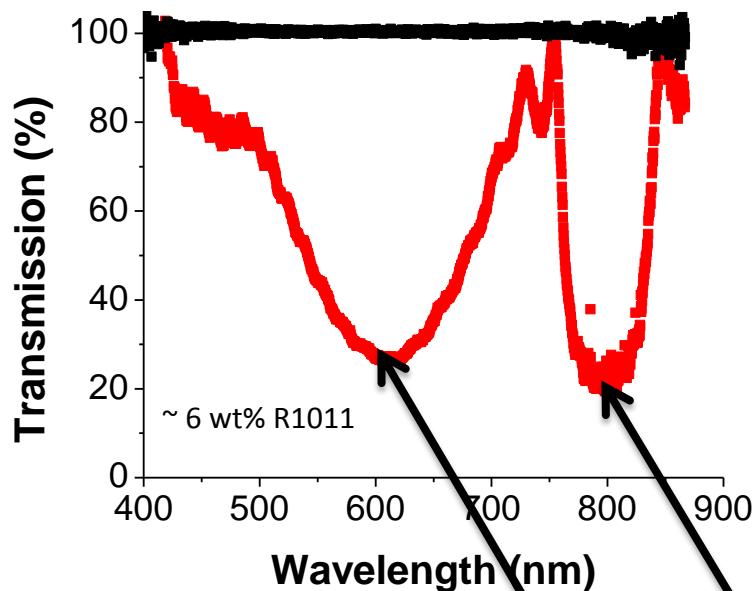


5CB



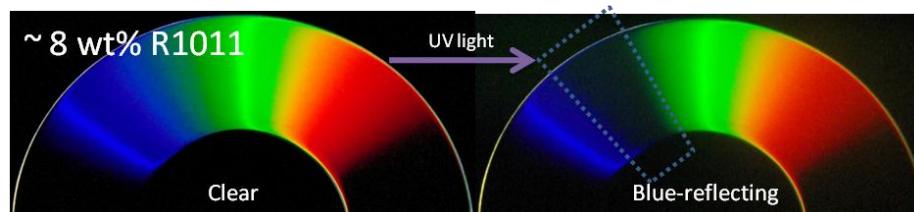
Data collected at AFRL/RX

Before irradiation – sample completely transmissive in VIS and NIR



After irradiation – sample becomes both absorptive and reflective

Data collected at AFRL/RX

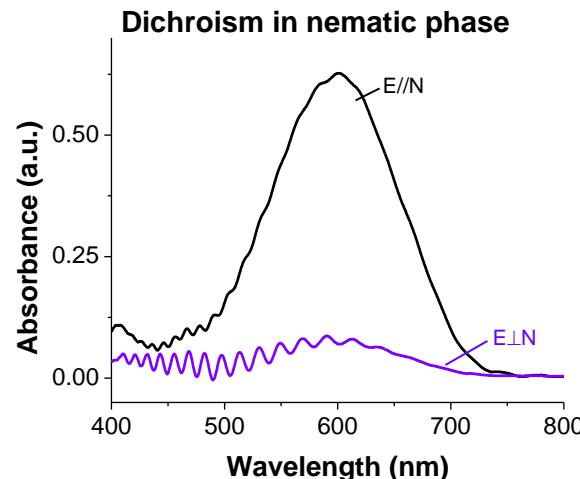
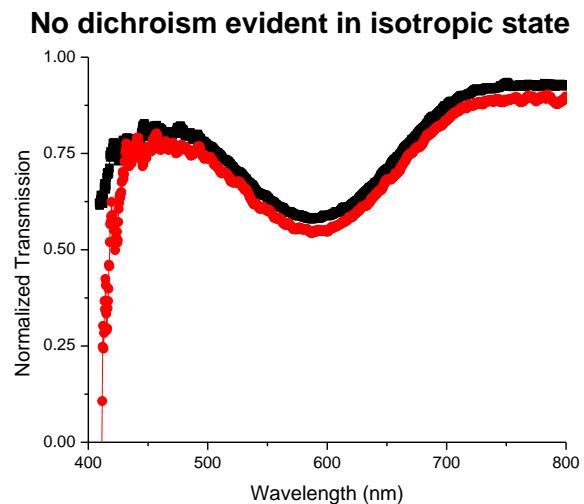




# Naphthopyran Phototropic Mixtures Unprecedented “Photo-dichroism”



For the Isotropic to Nematic Transition in  
AMI15/5CB Mixtures,  
Dramatic light induced changes in dichroic ratio  
from ~0 to 0.722



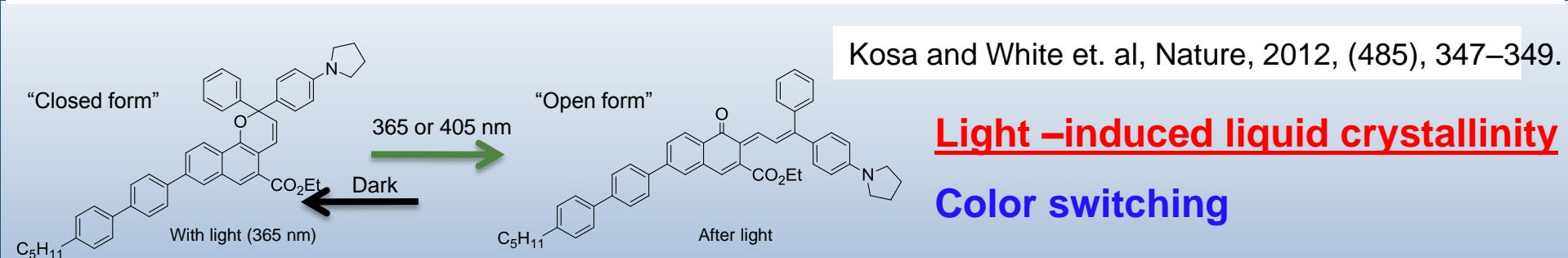
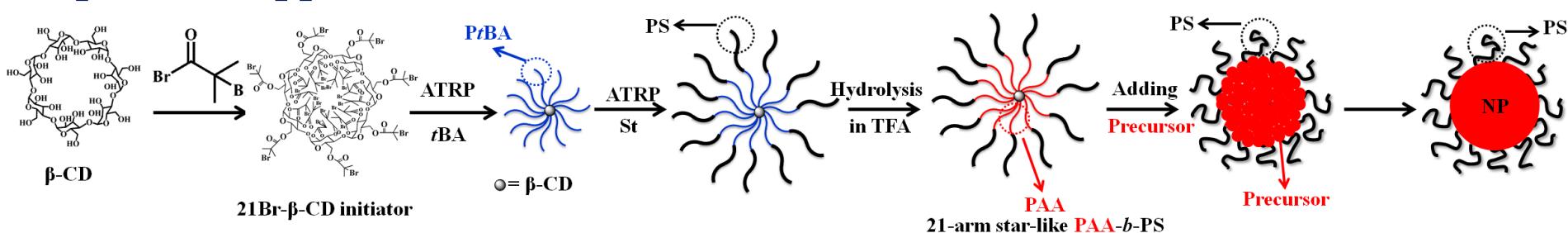
The mixture changes color and becomes polarized at the same time  
(Plain Glasses become Polarized Sunglasses)



# COE Georgia Tech/AFRL Joint Project



- To craft novel *organic-inorganic nanocomposites* composed of Superparamagnetic Iron Oxide Nanoparticles (SPION) *intimately and permanently connected* with nematic liquid crystals (LCs) and chiral azo molecules with high helical twisting power (HTP) for many potential applications.



potential for application in *communication devices, molecular devices, light-controllable devices, optical display system, optical data recording, photo-optical triggers, polarizers, and reflectors, and electromagnetic sensors, etc.*



# One-Dimensional Palladium Wires

## Tobias Ritter (YIP), Harvard U



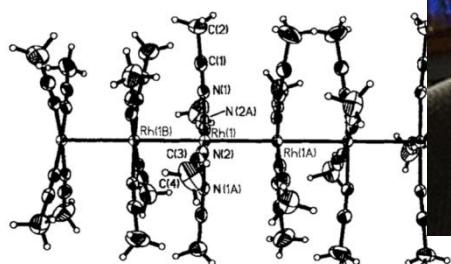
### Background on 1-D Metal Chains:

- Solid-state mixed-valence 1-D chains with Metal–Metal bonds
- Aqueous mixtures

-There are a few reports of infinite metal chains in the solid state with metal–metal bonds.

-Not solution stable; Solid-state

- take several days or weeks
- low yield (usually 50% or less)
- small scale (< 100 mg)



*Angew. Chem. Int. Ed.* **1969**, *8*, 35.  
*Angew. Chem. Int. Ed.* **1996**, *35*, 2772.  
*J. Organomet. Chem.* **2000**, *596*, 130.  
*Inorg. Chem. Commun.* **2001**, *4*, 19.



1-D metal chain  
characterized

mixed valence ( $d^7-d^8$ ) oligomers: Pt blues,  
Rh blues, Rh oligomers.

*Chem. Ber.* **1908**, *41*, 312.

*Science* **1982**, *218*, 1075.

*Coord. Chem. Rev.* **1999**, *182*, 263.

*Angew. Chem. Int. Ed.* **2001**, *40*, 4084.

*J. Am. Chem. Soc.* **1981**, *203*, 2220.

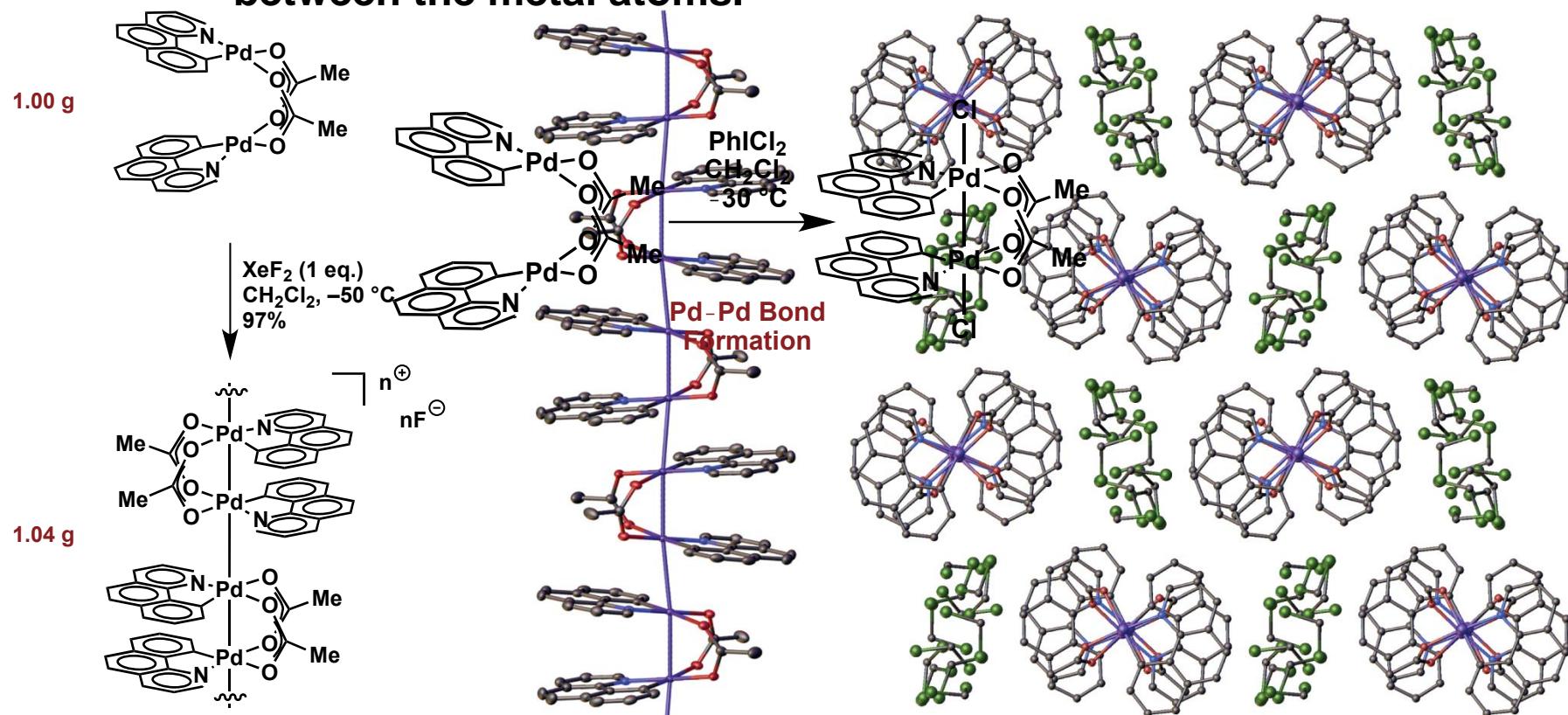


# New Chemistry – Solution Processible Palladium Wires



From Dimers to Wires:

- In Oxidative Addition chains in solution state revealed by X-ray crystallography
- Rapid, High-Yielding, Gram-Scale Solution-Phase Synthesis between the metal atoms.



The polymerization occurs in solution in less than 5 minutes, giving pure material on large scale

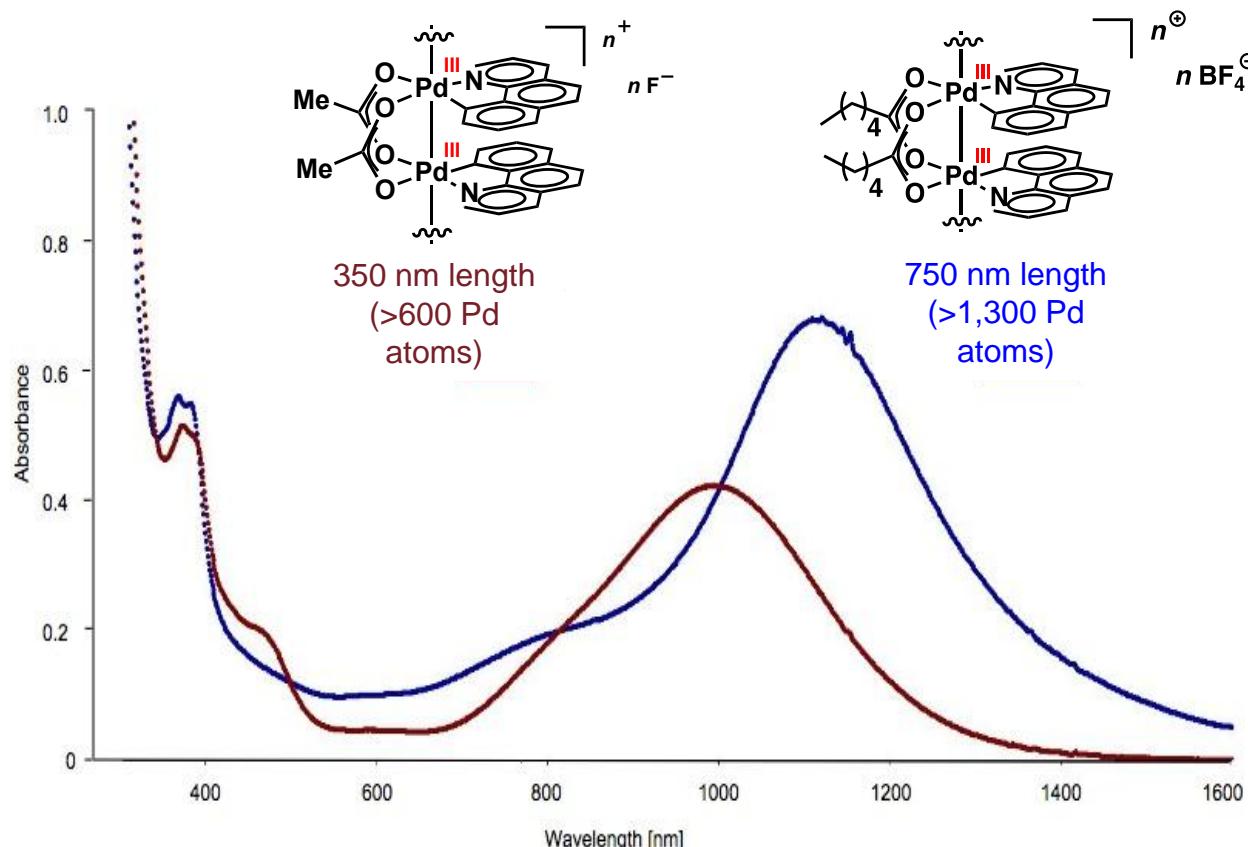


# Solution Stable 1-Dimensional Palladium Wire



1-D metal wires are predicted to display room temperature superconductivity

- Lengths up to 750 nm (**>1,300 Pd atoms**) observed in solution
- The longest solution-stable metal–metal bonded chain previously reported with assigned length contains **12 metal atoms<sup>‡</sup>**.
- Choice of counter-Anion controls chain length
- Enabled efficient device fabrication, not possible with previous 1-D wires



*Nature Chem.* 2011, 3, 949–953.  
†*J. Am. Chem. Soc.* 1981, 203, 2220–2225.



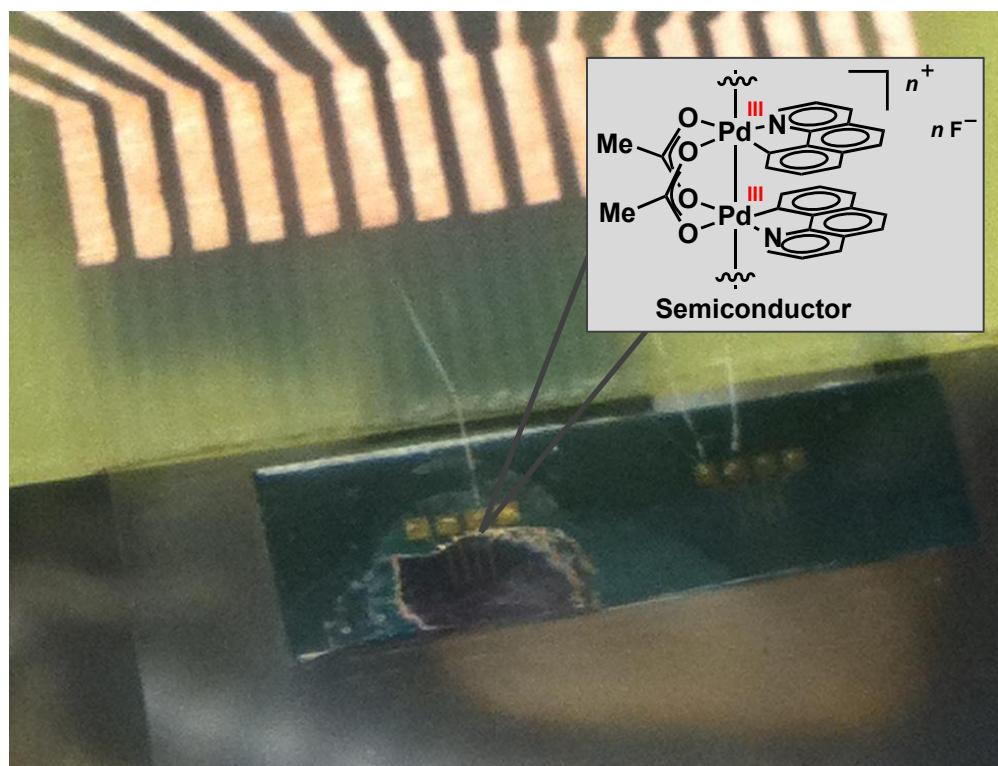
# Four Point Probe Measurement



## Thin-Film Conductivity:

- Solution processing capabilities allow for thin-film coating
- Four-point probe device used to measure conductivity of 1-D wire polymers film

Devices were fabricated using thin films of the 1-D wire polymers, which could be deposited from dichloromethane solutions either by drop casting or spin coating.



*Nature Chem.* 2011, 3, 949–953.

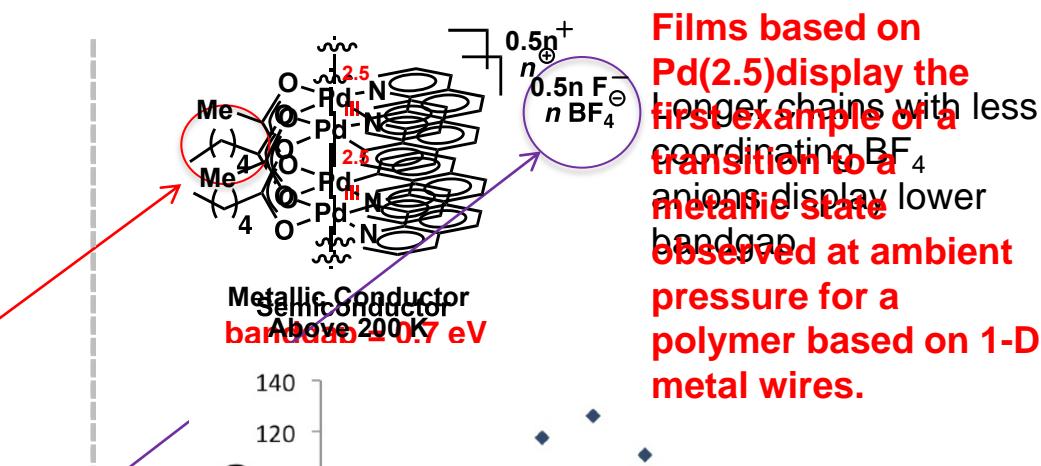
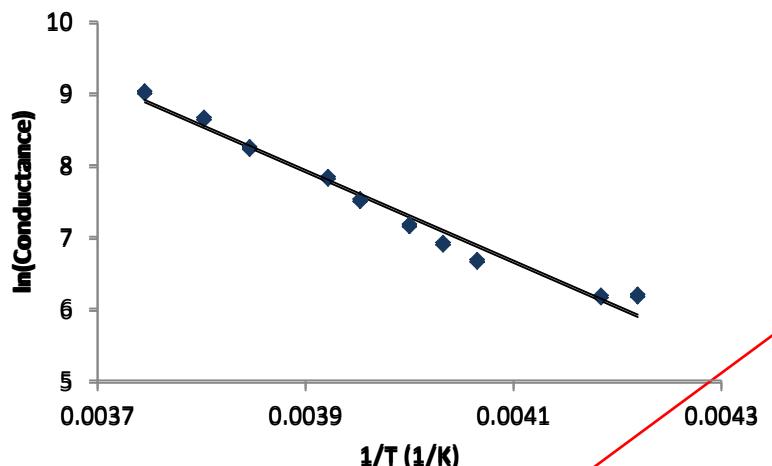


# Tuning of Electronic Properties



## Tuning Flexibility:

- Side Group Solubility
- Counter Ion
- Pd Oxidation State



Solution-stable 1-D metal wires with tunable conductive properties may have an impact on areas such as:

- Next-Generation Solar Cells
- Molecular Sensors
- Molecular Wires for Nanoscale Circuits

*Nature Chem.* 2011, 3, 949–953

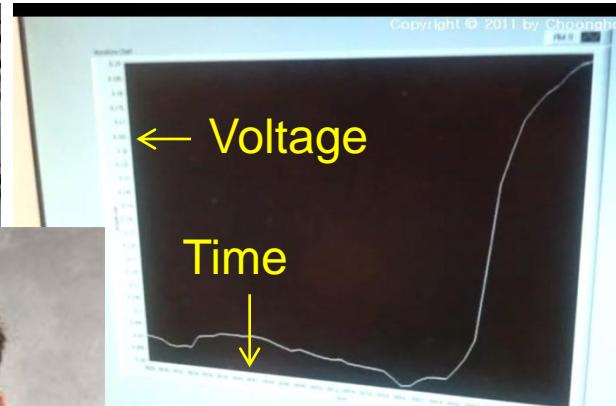
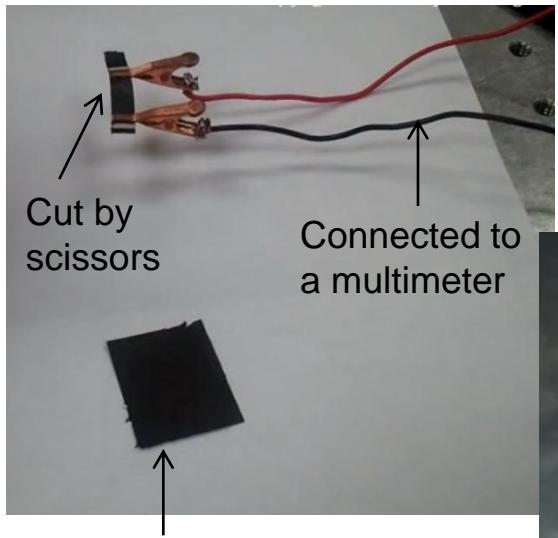


# Power Generation with Body Heat

Choongho Yu & Jaime Grunlan, Texas A&M

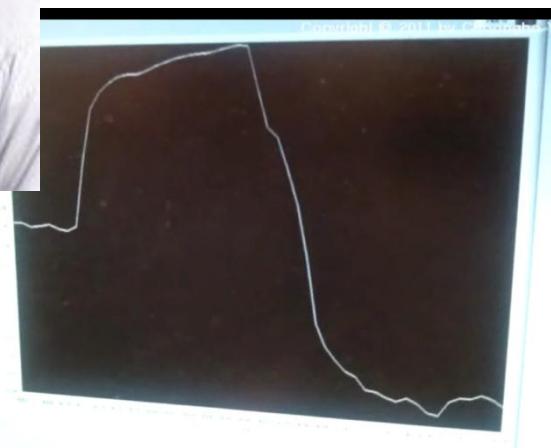
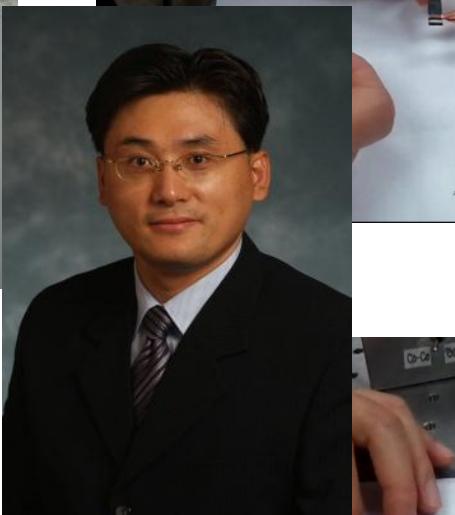
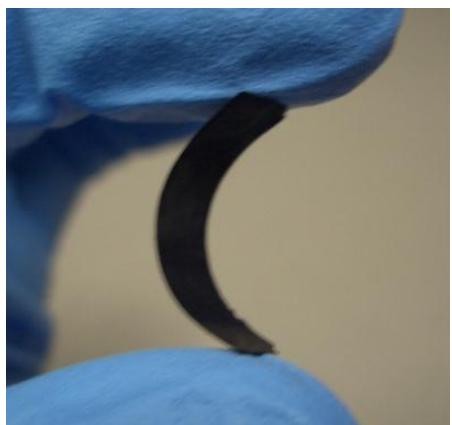


First demonstration of electricity generation from polymeric materials



Voltage –Time response

Flexible TE polymers



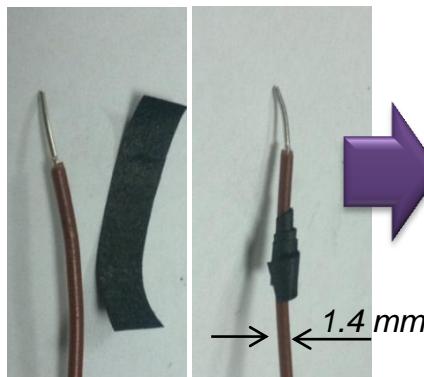


# Air-stable fabric thermoelectric modules made of n & p-type composites



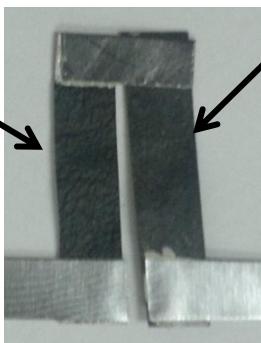
**Objective: Demonstrate power generation & cooling with organic composites**

## (1) Flexible composite



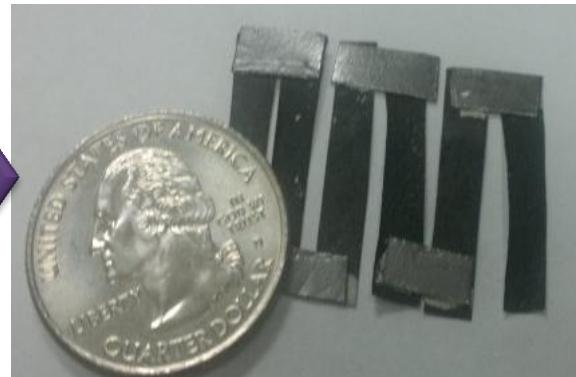
N-type  
*Carbon nanotubes + Poly-ethyleneimine (PEI) + NaBH<sub>4</sub> treatment*

## (2) Module fabrication

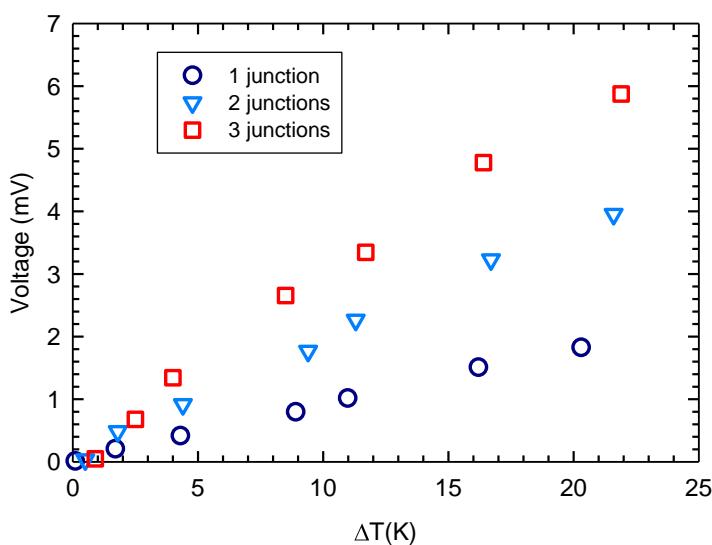


P-type  
*Carbon nanotubes + Paper (cellulose fibers)*

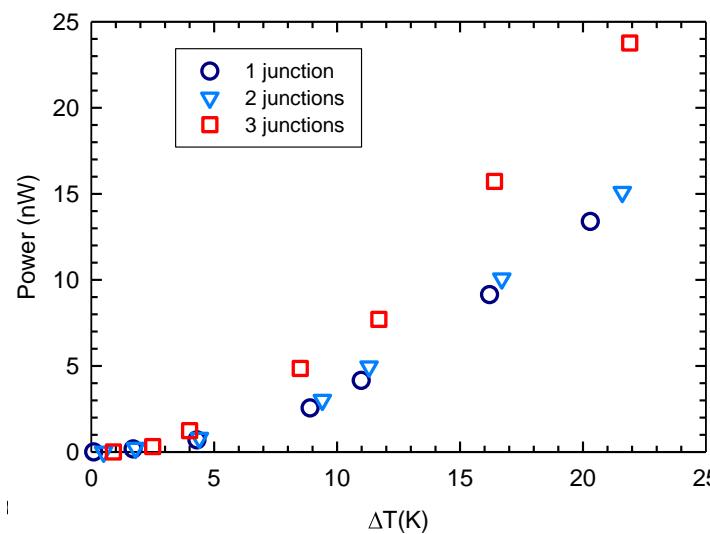
## (3) Multiple junctions in series



Voltage output vs Temperature



Power output vs Temperature



**Voltage and power are being increased by:**  
**(a) stacking more layers;**  
**(b) connecting more modules**



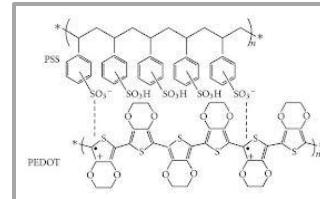
# Improving Power Factor by Tuning P-type composites with multiple CNT stabilizers



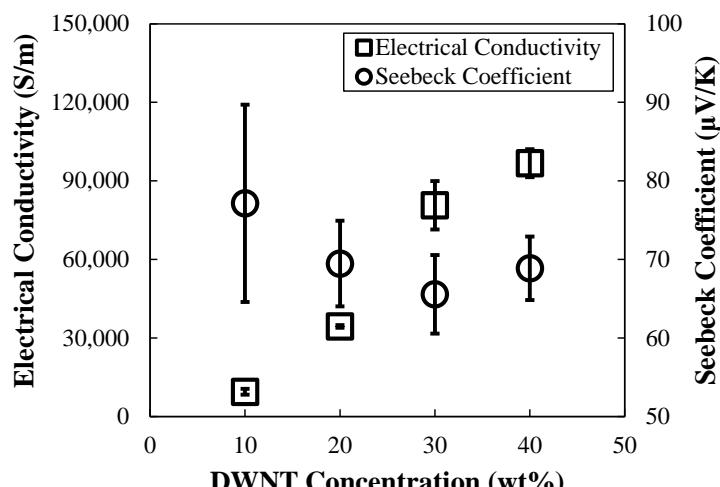
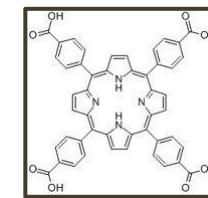
**Objective: Produce the highest possible power factor (PF) for fully organic, flexible composites**

- Double-walled carbon nanotubes (DWNT) are stabilized with two different molecules in poly(vinyl acetate) latex:

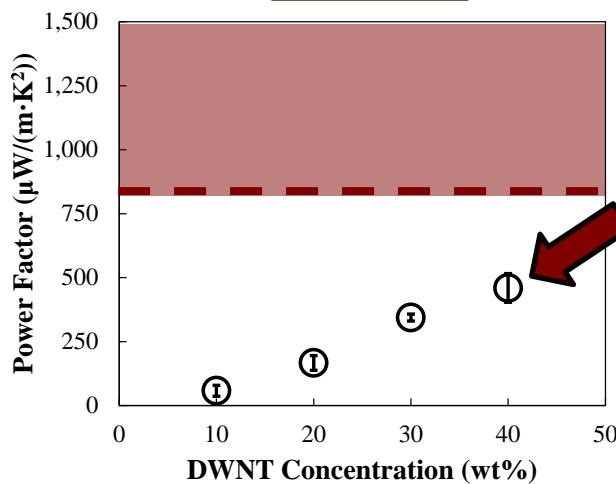
- PEDOT:PSS (conductive)



- TCPP (semi-conductive)



Electrical conductivity increases with DWNT concentration; while the Seebeck coefficient remains relatively insensitive.



Highest PF ever reported for fully organic composite at ~500  $\mu\text{W}(\text{m}\cdot\text{K}^2)$ !

The power factor ( $S^2\sigma$ ) increases with DWNT concentration and is within an order of magnitude of traditional inorganics (maroon shaded region).



# Different Module Design Concept

## David Carroll, Wake Forest U.



Using

s and

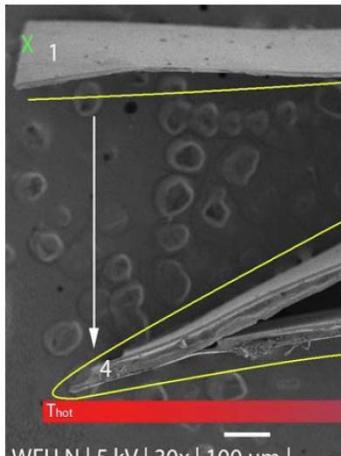


Figure 11: SEM image of multilayer conductive polymer films. The image shows a cross-section of four layers labeled 1 through 4. Layer 1 is p-type, and layer 4 is n-type. A red arrow at the bottom indicates the direction of current flow. A white arrow points to the right, and a green 'X' is at the top left. A scale bar of 100 µm is shown at the bottom right.



The garment has recently been shown on CNN International, CNBC, and the Discovery Channel.



# Photorefractive Polymers

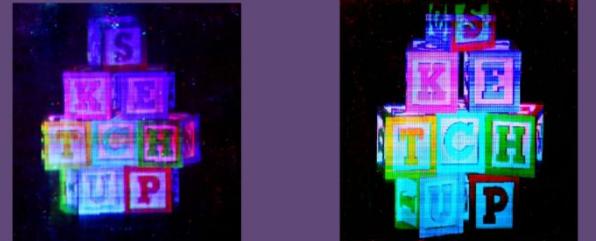
## Multi-TD's Interests



- Laser Refraction
- Optical Signal Processing
- Wave Front Correction
- 3D Holographic Display
- Image Correlation



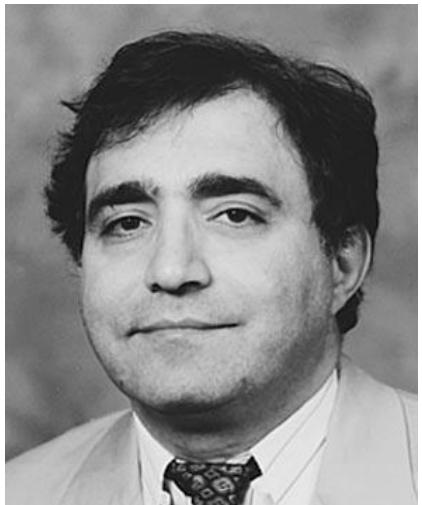
<u>Earlier Results</u>	<u>Now</u>
Luminance      350Cd/m <sup>2</sup>	1000Cd/m <sup>2</sup>
Image Holding    30s <50Cd/m <sup>2</sup>	2min >200Cd/m <sup>2</sup>
Sensitivity      200mW	1W



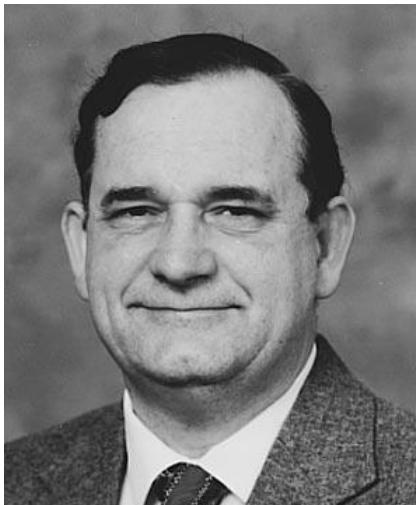


# Two Beam Coupling Optical Correlation

Jed Khoury AFRL/RY (11RY01COR)



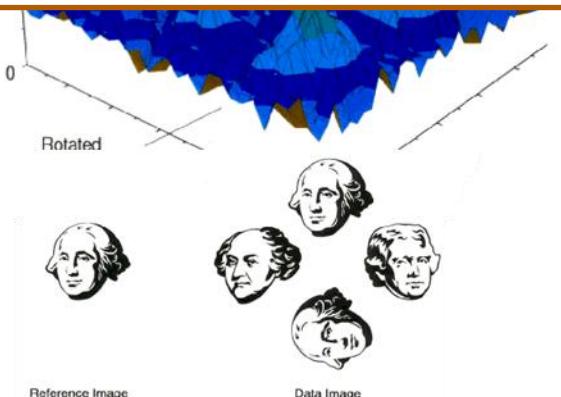
**Jed Khoury**



**Charles Woods**



**Bahareh Haji-saeed George Asimellis**  
**compression developed by**  
**AFRL/RY (Jed Khoury)**



2. **Organic photorefractive material that was developed by University of Arizona/Nitto Denko**

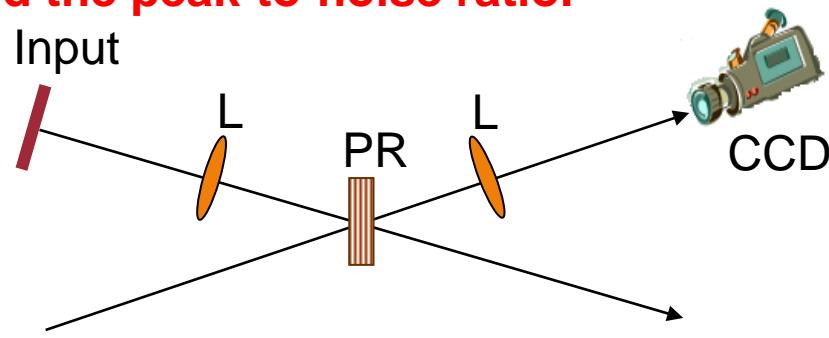
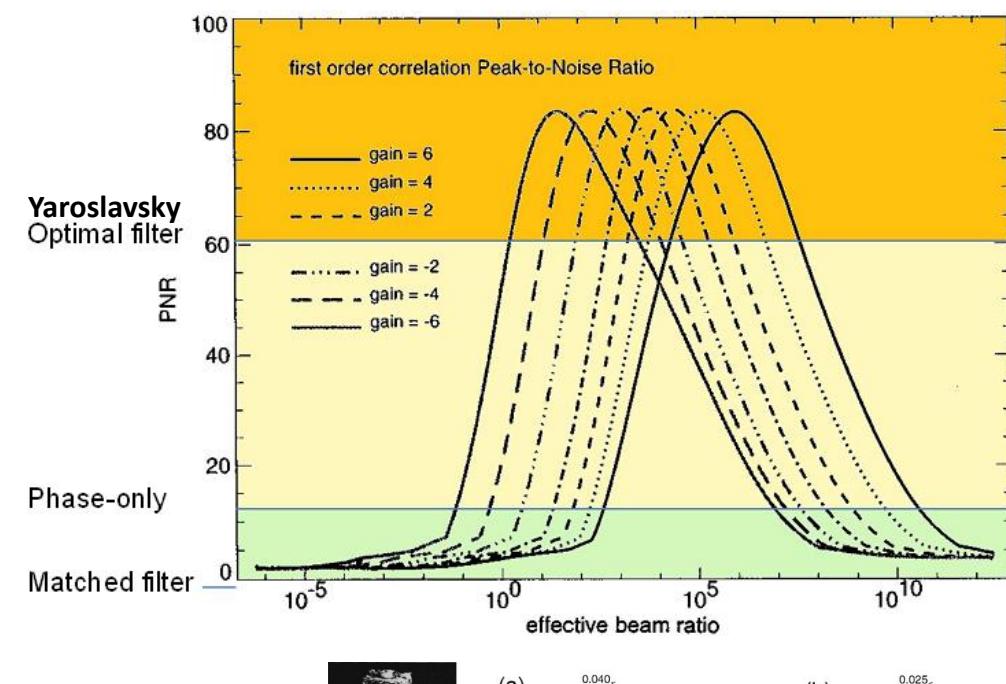
**Both efforts funded by AFOSR**



# Computer Simulation Comparing Two-Beam Coupling Correlation vs SOA Correlation Algorithms



No correlation filter in the last 50 years, since the first correlation invented by Vander Lugt (1963), have been designed that can improve simultaneously the discrimination, the signal-to-noise ratio, and the peak-to-noise ratio.



Using input that has a lot of background noise, Two Beam Coupling Correlation is:

- 1.5X better than Yaroslavsky Optimal filter
- 10X better than Phase-only filter
- superior to Matched filter (failed to recognize target)

But the scheme will require very large beam ratio, that will require a photorefractive material that has very high diffraction efficiency.



Input

Matched filter

Phase-only filter

Two Beam Coupling  
Compression filter





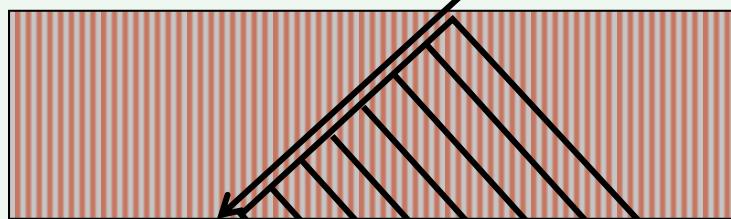
# BULK PHOTOREFRACTIVE CORRELATION VS THIN FILM PR POLYMER CORRELATION

Jed Khoury, AFRL/RY



A Thick BSO Crystal

Point source  
( $\delta$ -function input)



Thick diffracted beam  
(Broad impulse response )

A Thin Nitto Denko Organic Material

Point source  
( $\delta$ -function input)



Thin diffracted beam  
(Narrow impulse response )

Dephasing Factor is small in thin film holographic materials.

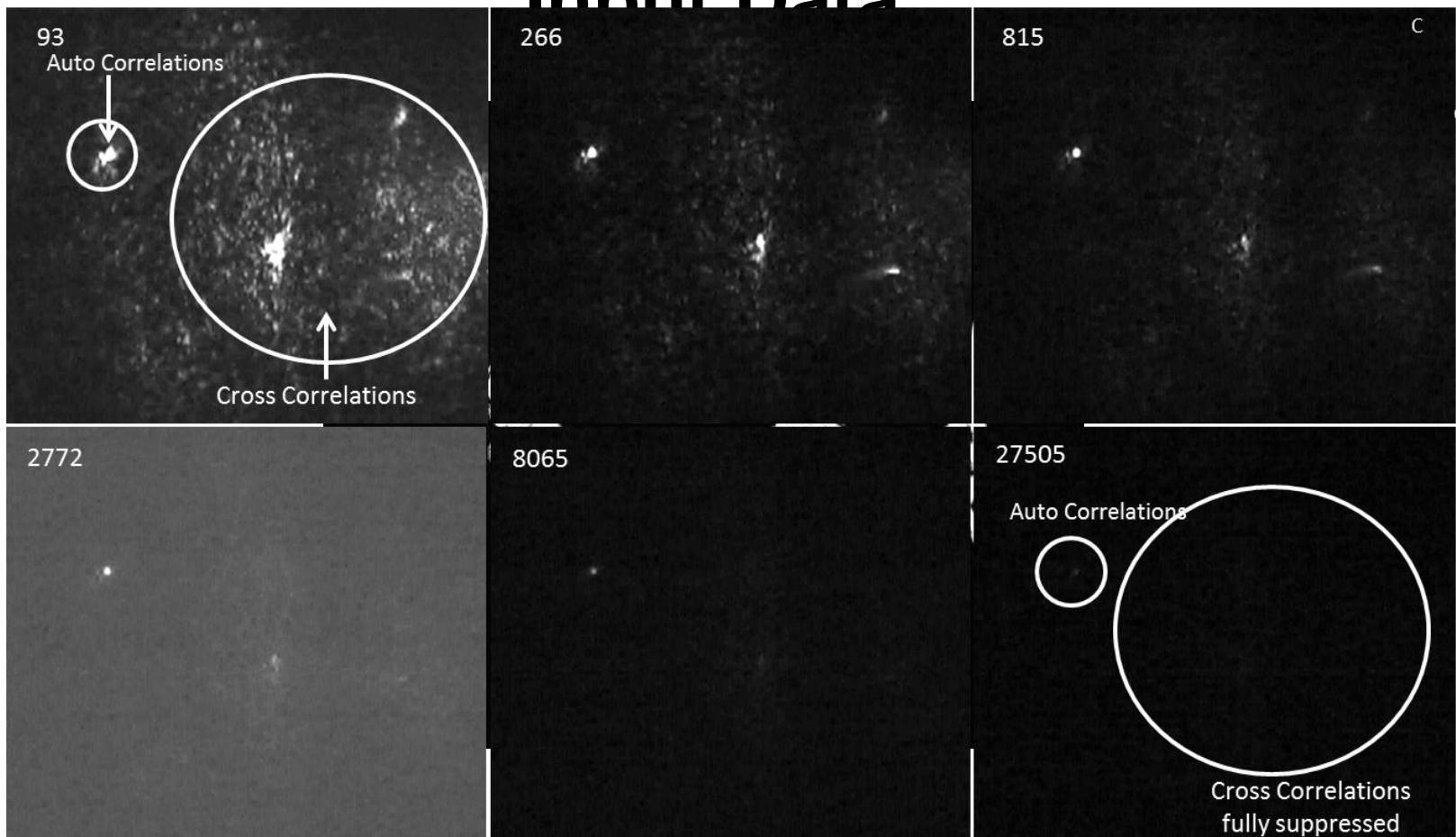


# Two Beam Coupling Experiment with PR Polymer Thin Film(1)



Dynamic range compression increases

## Input Data



Dynamic range compression increases

DISTRIBUTION STATEMENT A | Unclassified, Unlimited Distribution

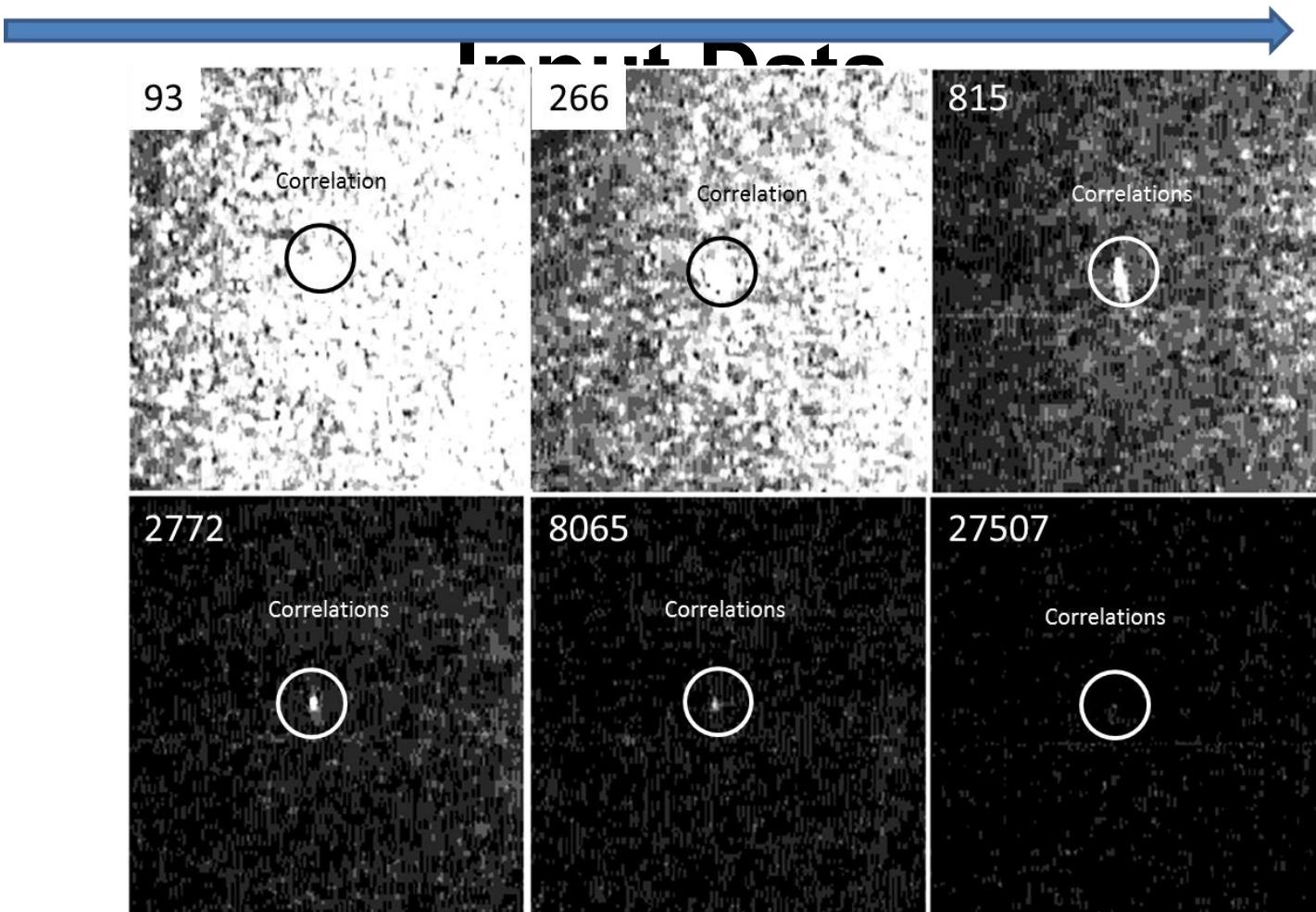




# Two Beam Coupling Experiment with PR Polymer Thin Film (2)



Dynamic range compression increases



Dynamic range compression increases

DISTRIBUTION STATEMENT A – Unclassified, Unlimited Distribution

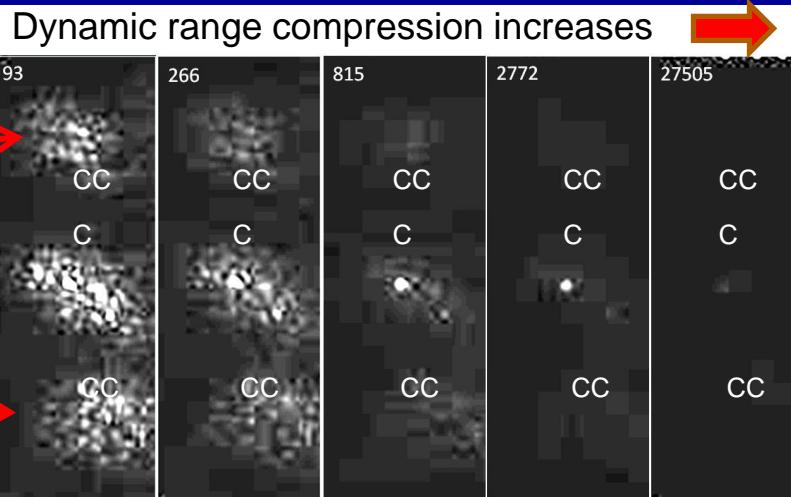
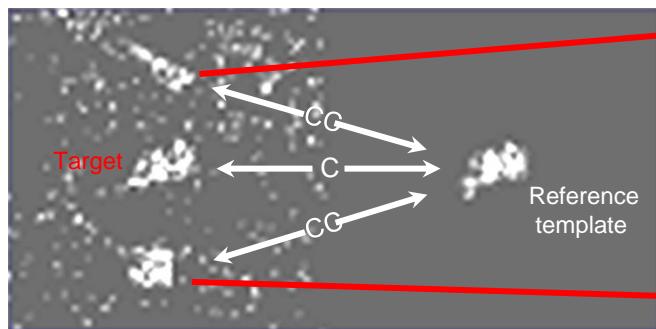




# Applied to Synthetic Aperture Radar Data



Low resolution images synthesized from the MSTAR data base



The first correlation filter that can improve simultaneously the

- SNR (100X)
- PNR,
- Discrimination (3 orders of Magnitude)

**Material Chemistry Makes It Possible!!!**

Correlation filter that outperforms optimal digital correlation filters



# Portfolio Trends

## Decreasing Emphases:

- Organic Solar Cells
- Organic Transistors

## Increasing Emphases:

- Self Assembly in Solid State
- Radical, Spin and Excited State Controlled Properties



# Summary



- Program Focused on developing New and Controlled Properties
- Not applications specific, but often use applications to guide the properties focuses
- Scientific Challenges
  - Discover New Properties
  - Control Properties
  - Balance Secondary Properties
- General Approaches
  - Molecular Design
  - Processing Control
  - Establish Structure Properties Relationship

